STEREOSELECTIVE CONVERSION OF LITHIATED BENZYLIC OR ALLYLIC

3-METHYL-1(Z),3-BUTADIENYL SULFIDES INTO CIS-DISUBSTITUTED CYCLOPROPANE COMPOUNDS

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Summary

The lithio-thiacyclohexenes 6 are intermediates in the title conversion $4 \rightarrow 9$.

In the course of our study of the nucleophilic trans-addition of thiols to 1-alkyn-3-ols ¹, we have shown that various 3-methyl-1(Z),3-butadienyl sulfides <u>3</u> are available through addition of thiols to 2-methyl-1-buten-3-yne in basic medium ². Thus the sulfide <u>3</u>a was obtained after reaction in dimethyl sulfoxide solution in the presence of 0.2 equiv. of potassium hydroxide powder at room temperature, in 75 % yield. Similarly, the thioacetate <u>1</u>b affords the (Z)-sulfide <u>3</u>b (62%) in the presence of 1.2 equiv. of sodium hydroxide.

Formation of a C-C bond between the methylene carbon adjacent to sulfur and the terminal carbon of the isoprenoid chain looked attractive due to the (Z) structure of compounds 3, the synthetic goal being the di-substituted thiacyclohexene 8 with R^1 , R^2 = regular terpenoid chains. Thus we first examined the properties of the lithic derivatives 4, expecting that these derivatives would give an intramolecular 1,4-addition to afford the cyclic carbanion 6³.

Addition of 1.1 equiv. of n-butyllithium (for $\underline{3}a$) or sec-butyllithium (for $\underline{3}b$) to a cooled solution (-78°) of the 1.3-dienes $\underline{3}$ in tetrahydrofuran resulted in rapid metalation (30 min)⁴. Subsequent reaction at - 78° with methyl iodide afforded the compounds ($\underline{5}a, b$). Quenching of the lithio derivatives ($\underline{4}a, b$) with diverse electrophiles was carried out under various conditions and the results are summarized in the table. At - 47°, the compound $\underline{4}b$ is stable, but $\underline{4}a$ is smoothly converted into the cyclic lithio derivative $\underline{6}a$. After stirring at - 30°, the lithio species did not give unique products when quenched with methyl iodide. The addition of one equiv. of hexamethylphosphoramide (HMPA) to the solutions of lithio derivatives ($\underline{4}a, b$) gave more rewarding results. Treatment with methyl iodide smoothly yielded the cyclopropane compounds (10a,b) charac-

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terized by the (Z) structure of the vinyl thioether group 5 and by the cis position of the two unsaturated side chains 6 .

Mild conditions for converting the lithium ene-thiolates <u>9</u> into the corresponding aldehydes <u>13</u> without destroying the fragile divinyl cyclopropane system were now sought. Quenching of the ene-thiolates <u>9</u> with ethylene oxide afforded the β -hydroxyethyl vinyl sulfides (<u>11</u>a,b) which were treated in chloroform with 0.1 equiv. of methanesulfonic acid at 0° during 10 min; the oxathiolannes (<u>12</u>a,b) were obtained with overall yields of 45 % and 42 % from (<u>3</u>a,b) ⁸. Final hydrolysis with methyl iodide in wet acetonitrile or with chloramine T in wet methanol ⁹ gave the aldehydes <u>13</u>a (75 %) and <u>13</u>b (51 %) ¹⁰.

Three features of this study are particularly noteworthy: (i) the conversion $4a \rightarrow 6a$ is a new case of the rare electrocyclisation reaction which was previously observed for lithiated C₁₀ isoprenoid sulfones ¹¹; (ii) the overall transformation $3 \rightarrow 12$ represents a new synthesis of cyclopropane compounds; (iii) the rearrangement of the intermediate lithic thiacyclohexenes $6 \rightarrow 9$ is highly stereoselective. The present state of our work does not allow us to propose a mechanism for this rearrangement. However it is possible that the favoured conformation of the carbanions <u>6</u> has the side chain R^1 oriented quasi equatorially for steric reasons, as it was already discussed for the rearrangement of lithiated nerol oxide ¹⁰ and this would account for the stereoselectivity observed.

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Substrate	Conditions ⁱ	Yield % ⁱⁱ	<u>5</u>	<u>Z</u>	Products <u>8</u> (trans:cis)	<u>10</u>	<u>11</u>
<u>3</u> a	Quenching at -78° with CH ₃ I	88	100	-	_	-	-
<u>3</u> b	idem	80	100	-	-	-	-
<u>3</u> a	Warming to -47° then 2 h at -47°; quenching with CH ₃ I	75	-	-	100 (72/28)	-	-
<u>3</u> b	idem	(93)	100	-	-	-	-
	Warming to -30° then 2 h at -30°; quenching with						
<u>3</u> a	H ₂ 0	50^{iii}	-	100	-	-	-
<u>3</u> a	сңзі	(92)	-	-	4 1 (71/29)	59	-
<u>3</u> b	СНЗІ	(95)	25	-	5	70	-
	l equiv HMPA added, warming to -30°then 2 h at -30° and quenching with						
<u>3</u> a	СН3І	70	-	-	_	100	_
<u>3</u> b	CH3I	72	-	-	-	100 ⁱ '	, _
<u>3</u> a	ethylene oxide	(85)	-	-	-	-	100
<u>3</u> b	ethylene oxide	(71)	-	-	-	-	100

i In all cases, l.l equiv. of n.BuLi (for $\underline{3}a$) or of sec.BuLi (for $\underline{3}b$) was added to a cooled (-78°) solution of the 1,3-diene $\underline{3}$ in THF and the mixture stirred at that temperature for 30 min.

ii The numbers in brackets are the yields of crude products.

iii The lithium thiolate <u>9</u>a is eliminated by washing the organic phase with water during the work up.

iv ca 4 % of the trans isomer can be detected in the ¹H-N.M.R. spectrum.

REFERENCES AND NOTES

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